

Worksheets: Buffers, Titration

1. Give an example of a buffer solution. Why is it important to have both an acid and base present?

2. Why is titration used?

3. During a titration experiment, it took 72 mL of 0.500M H_2SO_4 to neutralize 321mL of a solution of KOH. What is the concentration of the KOH solution?